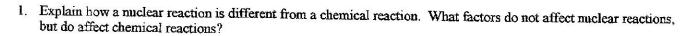
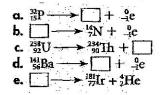
## Honors Chemistry: Homework Set 3-1



- 2. Compare and contrast the three main types of nuclear radiation.
- 3. How is the atomic number of a nucleus changed by alpha decay? By beta decay? By gamma decay?
- 4. How does alpha decay affect the mass number of a nucleus? How does beta decay affect the mass number?
- 5. When polonium-210 decay by alpha radiation, what isotope is formed?
- 6. Complete these nuclear equations.



- Alpha radiation is emitted during the decay of the following isotopes. Write balanced nuclear equations to describe each decay process.
  - **a.** 231 Pa **b.** 241 Am c. 226 Ra **d.** 252 Es
- 8. The following radioisotopes are beta emitters. Write balanced nuclear equations to describe each decay process.
  - a. <sup>3</sup>H b. <sup>28</sup>Mg c. <sup>13</sup>H d. <sup>75</sup>Se